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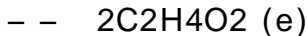
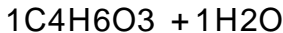
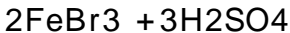
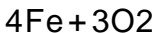
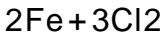
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working hard and
surely you will
succeed.

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(Coefficients equal to one (1) do not need to be shown in your answers). (a)



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(f) $1\text{C}_4\text{H}_{10}\text{O} + 6\text{O}_2$

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– – $4\text{CO}_2 + 5\text{H}_2\text{O}$

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1) $2 \text{NaNO}_3 + \text{PbO}$

$\text{Pb}(\text{NO}_3)_2 + \text{Na}_2\text{O}$

2) $6 \text{AgI} + \text{Fe}_2(\text{CO}_3)_3$

$2 \text{FeI}_3 + 3 \text{Ag}_2\text{CO}_3$

3) $\text{C}_2\text{H}_4\text{O}_2$

$+ 2 \text{O}_2 \rightarrow 2 \text{CO}_2 + 2$

H_2O

4) $\text{ZnSO}_4 +$

$\text{Li}_2\text{CO}_3 \rightarrow \text{ZnCO}_3 +$

Li_2SO_4

5) V_2O_5

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+ 5 CaS 5 CaO + V

2S5 6) Mn(NO₂)₂

+ BeCl₂ 2 Be(NO

2)₂ + MnCl₂ 7) 3

AgBr + GaPO₄ 4 Ag

3PO₄ + GaBr₃ 8)

3 H₂SO₄ + 2

B(OH)₃ B₂(SO₄)₃

+ 6 H₂O 9) S₈ + 8

O₂ 8 SO₂

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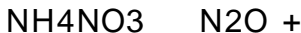
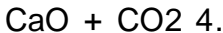
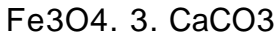
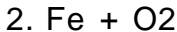
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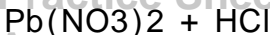
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$Al_2(SO_4)_3 + NaOH$
 $\rightarrow Na_2SO_4 + Al$

$(OH)_3$. 3. $Al_2O_3 +$
 $Fe \rightarrow Fe_3O_4 + Al$. 4.
 $KClO_3 \rightarrow KCl + O_2$.

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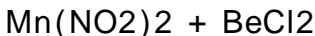
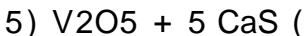
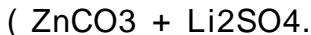
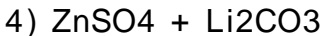
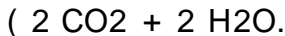
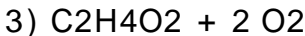
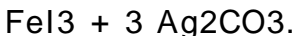
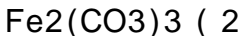
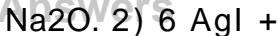
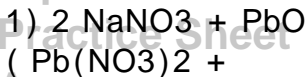
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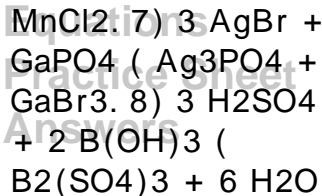
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2(SOFe₄)₃ + 3 .

H₂ . 2 . C . 2 . H₂ . 62

+ 7 . O₂ . 6 . 2O .

H₂ . 4 . CO . 2 . 3 .

KOH . 3 + 1 . H .

3PO₄ . 1 . 3PO₄K

+ 3 . H₂O . 4 . SnO

21 + 2 . H₂ . 1 . Sn

+ 2 . H₂O . 5 . NH

3 . 4 + 5 . O . 2 . 4 .

NO + 6 . H₂O . 6 .

KNO₃ + 1 . H

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2CO₃ · 1. 2CO₃ +
2 · HNO₃ · 3 · 7 · B · 2 ·
Br · 61 + 6 · HNO₃ · 3 ·
2 · B(NO₃)₃ + 6 ·
HBr · 8 · BF₃ · 2 +
3 · Li₂SO₃ · 1 ·
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